

Figure 1 (a): SEM micrographs of unmodified (CCM) sample

Fig. 1 (b) shows FE-SEM image of (0.5% TNP + 0.4 % WPF). The image showed a large amount of gel like materials existence, which is C-S-H that is responsible to fill the pores in the concrete microstructures to enhance the concrete properties. Uniformly dispersed TiO₂ nanoparticles accelerated the formations of C-S-H components in the whole concrete structure by controlling the crystallization process and hence, restricted the growth of Ca(OH)₂. TiO₂ nanoparticle serves as the nuclei, which will change the hydration process, and thus, leading to the morphology clange of the hydration products. Furthermore, it could be the to fill the nanosized pores in the anorphous cement flous hydration products because permeability of aggressive environmental conditions, and tend to increase density C-S-H phase.

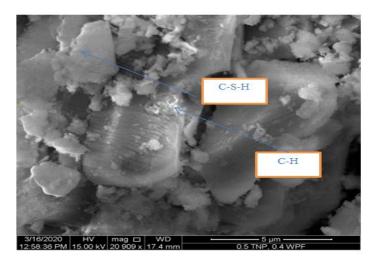


Figure 1 b: SEM micrograph of 0.5 % TNP + 0.4 % WPF modified sample

FE-SEM micrograph of 1 % TNP modified concrete shown in Figure 1 (c) indicated large amount of gel like material, which is C-S-H generated. At the same time, crystallization of concrete components is very limited. The partial replacement of cement with TNP accelerated

the hydration process in concrete, and helped to generate more C-S-H gel like material, which is responsible for improving durability and mechanical properties.

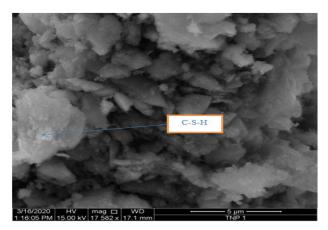


Figure 1 (c): SEM micrograph of 1% TNP modified concrete

Figure 2 shows the XRD results for the CCM and concrete modified with TiO_2 nanoparticles and waste plastic fiber. In case of CCM, the samples have shown unhydrated C-H phases at $(2\Theta = 10.90^{\circ}, 29.44^{\circ}, \text{ and } 29.46^{\circ})$. However, these phases were suppressed (absent) for the modified (1 % TNP and 0.5 % TNP and 0.4 % WPF) samples. Furthermore, the CCM samples have the unhydrated C_3S and C_2S phases at $(2\Theta = 26.72^{\circ})$ and $(2\Theta = 28.76^{\circ})$, respectively. In case of TiO_2 nanoparticles modified samples, the hydrated C-S-H phases appeared at $(2\Theta = 47.14^{\circ})$ and $(2\Theta = 47.14^{\circ})$ and the C-H phases that appeared in CCM sample at $(2\Theta = 10.90^{\circ})$ was possibly converted into C-S-H phase in both (0.5 %) TNP & 0.4 % WPF) and (1%) TNP) samples.

The SiO₂ (quartz) compositions peaks appeared for all the diffractograms at $(2\Theta = 18.17^{\circ}, 29.91^{\circ}, \text{ and } 31.18^{\circ})$. The modifications of concrete materials with the partial replacement of cement with TNP and sand with WPF, ensured with the increase in the peak corresponding to C-S-H phases, and simultaneously reduction in the C-H, C₃S and C₂S phases. This could be attributed due to the enabling of hydration reactions in the cement paste, and the consumption of $(CaOH)_2$, C₂S and C₃S to form an extra C-S-H gels like material.

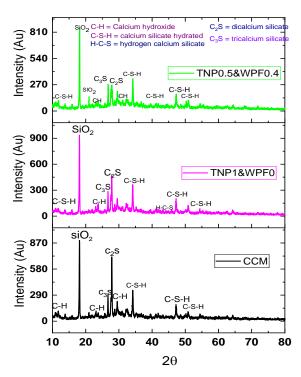


Figure 2: XRD spectra of unmodified (CCM), and modified with 1% TNP and (0.5% TNP & 0.4% WPF) modified samples

(Figure 3), the thermo-gravimetric tests were also done to find the thermal stability of concrete materials. The TGA-DSC tests were conducted starting from room temperature to 800°C temperature ranges with increasing heating rate of 10°C/minute. The weight loss has been appeared up to about 100 °C was due to dehydration and moisture removal from the sample. The maximum weight loss has recorded at the temperature of 426.43 °C and 433.85 °C for CCM; and TNP and WFP modified samples due to dehydroxylation and other hydration products, respectively. The mass loss occurred at about 668°C was attributed to the decarbonation of calcite (CaCO₃) in cement paste.

The DSC plot exhibited three main exothermic events during the entire heating process. The first occurs on the temperatures range from the onset to 77 °C for the unmodified sample, and at 82 °C for the modified samples. These weight losses were resulted from the dehydration and evaporation of moisture contents. Similarly, for modified concrete, (fig 3.6b), exhibits three main exothermic events. The first exothermic reaction occurred from onset temperature to 106 °C. At 106 °C, heat is absorbed - endothermic transition. From 106 °C to 380 °C, crystallization occurred - an exothermic event. Finally, bond formations occurred. At temperature of 380 °C again melting take place. Furthermore, from 380 °C to 600 °C crystallization undertakes.

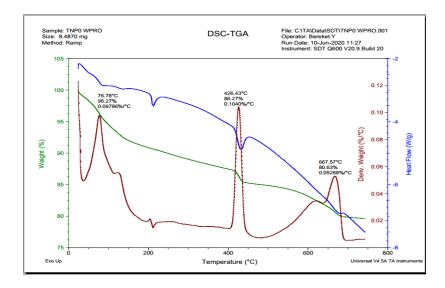


Figure 3a: DSC-TGA result of unmodified concrete materials

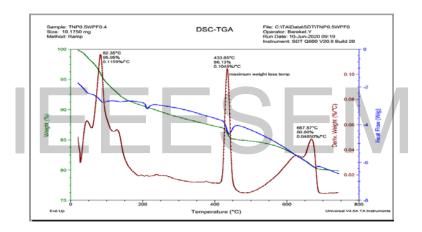


Figure 3 b: DSC-TGA results of 0.5 % TNP & 0.4 % WPF modified concrete

Conclusions

In this paper, the combined effects of TiO₂ nanoparticles and waste plastic fiber with various compositions were investigated. The results from mechanical tests indicated that both TiO₂ nanoparticles and waste plastics fibers have improved the mechanical properties like compressive strength, splitting tensile strength and maximum load resistance of the concrete materials. In addition, all of the modified concrete showed enhanced properties with respect to water permeability. In particular, 0.5 % TNP + 0.4 % WFP sample showed remarkable improvement in all aspects. For instance, the compressive strength of both TiO₂ nanoparticles and waste plastic fibers modified concrete after 28 days of curing time showed 29.01 to 39.05 MPa. Similarly, the split tensile strength showed increment from 5.14 to 9.52 MPa.

From the morphological study; the microstructure of modified concrete showed well-formed gels like materials, this is the indication of the existence of C-S-H. The thermal analysis of modified concrete indicated weight loss appeared at relatively higher temperatures than unmodified concrete sample. Furthermore, the XRD test result of unmodified concrete has shown the existence of various unhydrated phases like C-H, C₂S, C₃S and other unhydrated phases. Contrarily, the presence of hydrated phases, such as C-S-H gel like phases have confirmed modification in concrete materials. In general, the concrete materials modified with TiO₂ nanoparticles and waste plastics fiber has clearly shown developed cracks-resistance behavior in concrete structure.

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